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1.1 Molecular Mass Transfer For a binary mixture of A and B, the molar flux, $N_{A,z}$, of species A relative to the z axis is $N_{A,z} = -cD_{AB} \frac{dy_A}{dz} + y_A(N_{A,z} + N_{B,z})$ (1.1) In this equation, c is the total molar concentration, D_{AB} is the diffusivity of A in B, y_A is the mole fraction of A, and $N_{B,z}$ is the molar flux of B.

Chapter 1 Fundamentals of Mass Transfer

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mass transfer will be used throughout this module. 10.1 Mass transfer through diffusion. In Module 2 "Conduction", the Fourier equation was introduced, which relates the heat transfer to an existent temperature gradient: $q = -k \frac{dT}{dy}$ (Fourier's law) (10.1) For mass transfer, where a component A diffuses in a mixture with a component

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The principles of mass transfer, both diffusional and convective have been comprehensively discussed. The application of these principles to separation processes is explained. The more common separation processes used in the chemical industries are individually described in separate chapters.

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